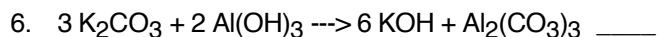
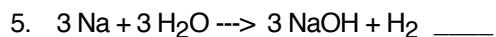
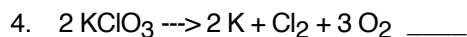
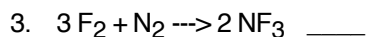
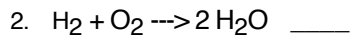
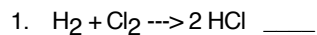
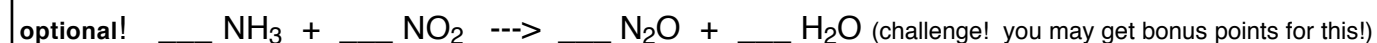
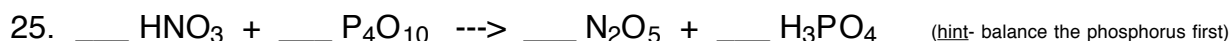
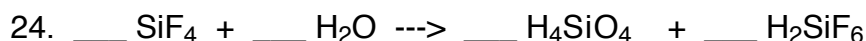
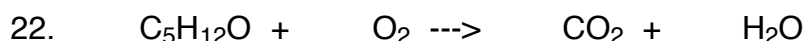
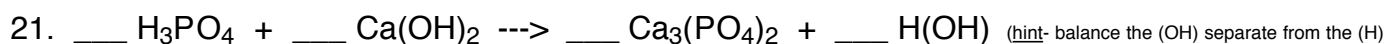
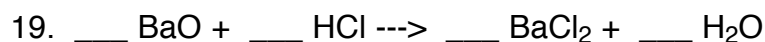
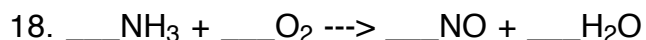
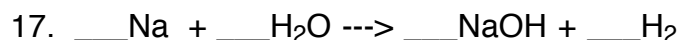
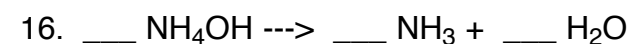
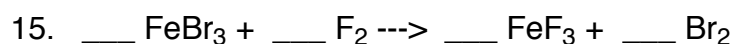
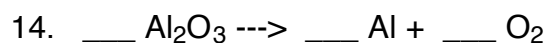
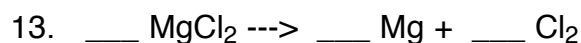
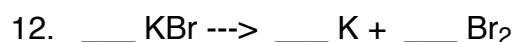


## WS 4.1 Balancing Equations / Formula Mass

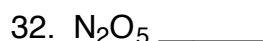
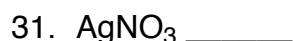
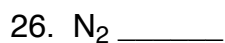
For 1-6, take inventory of each side and determine whether the equation is balanced (Y) or not (N):



For 7-25, balance the equation by writing in the appropriate coefficients (lowest whole-numbers). Check your answers by taking inventory (like above). **HINT:** use a pencil or erasable pen!!!!!!



For #26 - 33, Use a periodic table to determine the formula mass (atomic weight) of the following: use ans. bank...



Ans (IRO+2) (#26-33): 28 18 44 60 74 108 132 170 194 339